



## Amino Acids Found in Proteins - Part 1

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The common amino acids are grouped according to whether their side chains

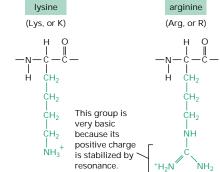
> acidic basic

uncharged polar nonpolar

These 20 amino acids are given both three-letter and one-letter abbreviations.

Thus: alanine = Ala = A

## BASIC SIDE CHAINS



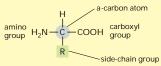
(His, or H)

histidine

These nitrogens have a relatively weak affinity for an H<sup>+</sup> and are only partly positive at neutral pH.

#### THE AMINO ACID

The general formula of an amino acid is

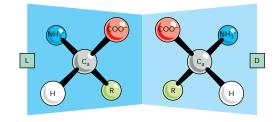


R is commonly one of 20 different side chains. At pH 7 both the amino and carboxyl groups are ionized.

$$H_3N - C - COO$$

### **OPTICAL ISOMERS**

The a-carbon atom is asymmetric, which allows for two mirror image (or stereo-) isomers, L and D.



Proteins consist exclusively of L-amino acids.

## PEPTIDE BONDS

Amino acids are commonly joined together by an amide linkage, called a peptide bond.

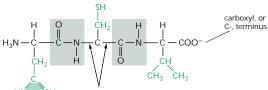
Peptide bond: The four atoms in each gray box form a rigid planar unit. There is no rotation around the C-N bond.

**Proteins** are long polymers of amino acids linked by peptide bonds, and they are always written with the N-terminus toward the left. The sequence of this tripeptide is histidine-cysteine-valine.

amino, or



= NH



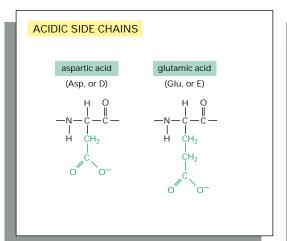
These two single bonds allow rotation, so that long chains of amino acids are very flexible.

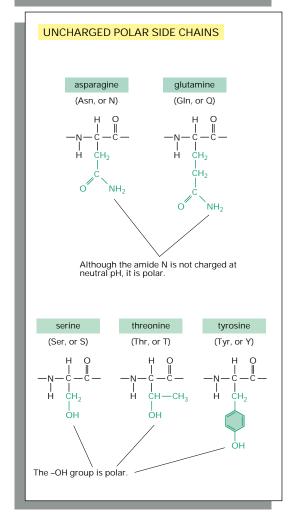


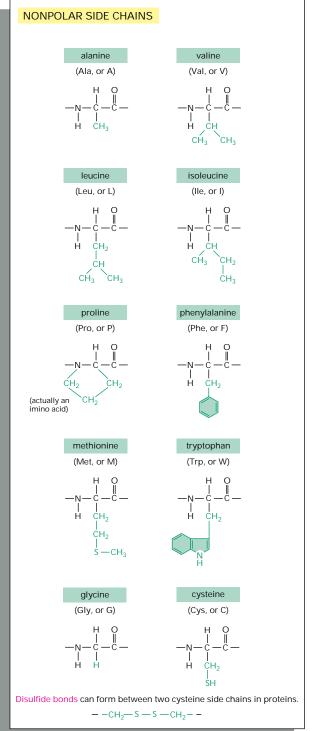


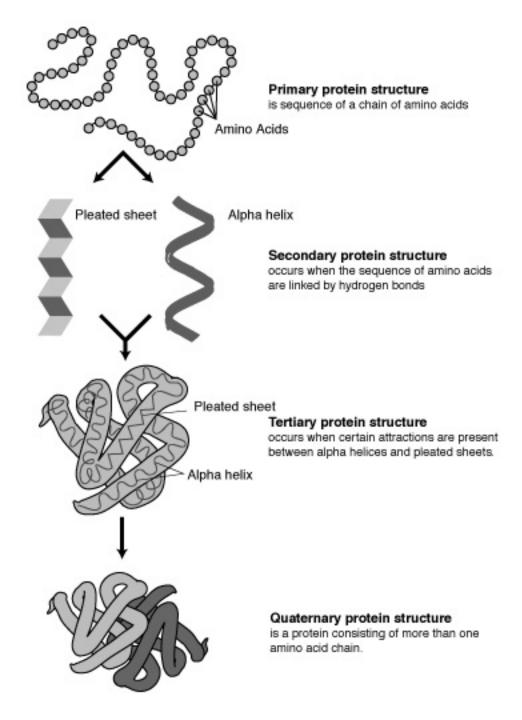
# **Amino Acids Found in Proteins - Part 2**

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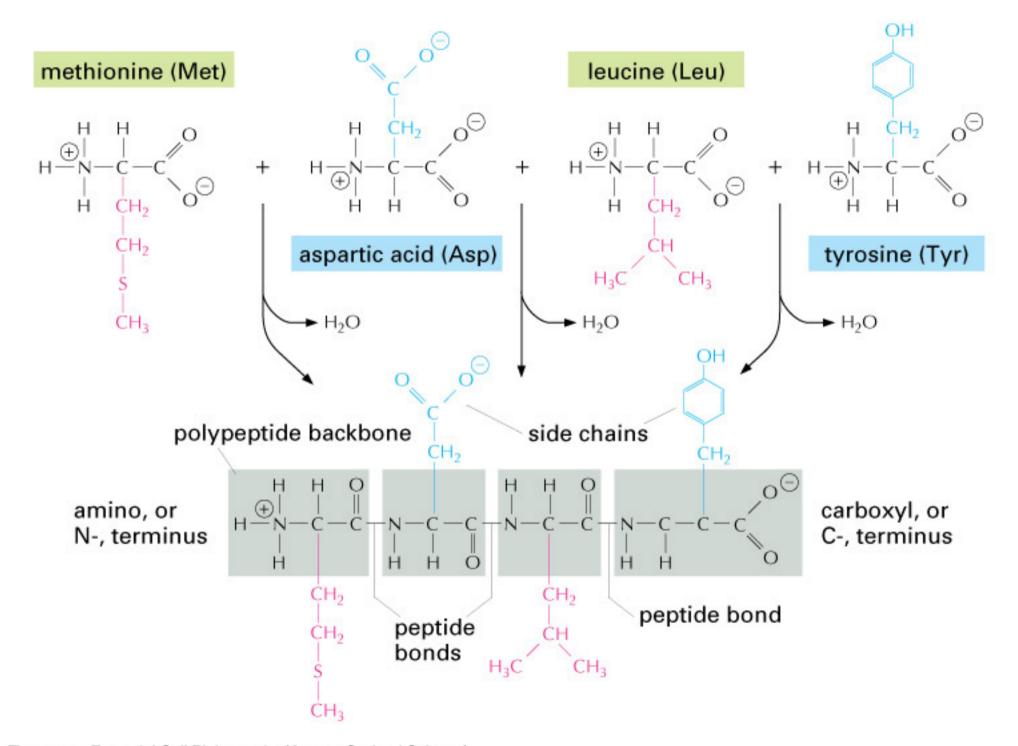


Figure 4-2 Essential Cell Biology, 2/e. (© 2004 Garland Science)